# MARINE COATING PERFORMANCE A SIX YEAR REPORT

October 1985

Prepared by
Associated Coating Consultants
Galveston, Texas 77551
in cooperation with
Avondale Shipyards, Inc.
New Orleans, LA 70150

a. REPORT unclassified	b. ABSTRACT unclassified	c. THIS PAGE unclassified	SAR	OF PAGES 39		
16. SECURITY CLASSIFICATION OF:  17. LIMITATION OF ABSTRACT					19a. NAME OF RESPONSIBLE PERSON	
15. SUBJECT TERMS						
14. ABSTRACT						
13. SUPPLEMENTARY NO	OTES					
12. DISTRIBUTION/AVAILAPPROVED for publ	LABILITY STATEMENT ic release, distributi	on unlimited				
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)		
9. SPONSORING/MONITO	PRING AGENCY NAME(S) A	AND ADDRESS(ES)		10. SPONSOR/MONITOR'S ACRONYM(S)		
Naval Surface War	IZATION NAME(S) AND AE rfare Center CD Co n 128 9500 MacArth	0	8. PERFORMING ORGANIZATION REPORT NUMBER			
				5f. WORK UNIT NUMBER		
				5e. TASK NUMI	BER	
6. AUTHOR(S)				5d. PROJECT NU	JMBER	
				5c. PROGRAM ELEMENT NUMBER		
<b>Marine Coating Pe</b>	erformance A Six Ye	ear Report		5b. GRANT NUN	MBER	
4. TITLE AND SUBTITLE				5a. CONTRACT	NUMBER	
1. REPORT DATE OCT 1985	2. REPORT TYPE <b>N/A</b>			3. DATES COVERED		
including suggestions for reducing	this burden, to Washington Headquuld be aware that notwithstanding ar		ormation Operations and Reports	, 1215 Jefferson Davis	Highway, Suite 1204, Arlington	

Public reporting burden for the collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and

**Report Documentation Page** 

Form Approved OMB No. 0704-0188

# Table of Contents

Table of Contents	Page 2
List of Figures	3
List of Tables	4
Foreword	5
Executive Summary	6
1.1 Project Overview 1.2 Cost Savings 1.3 Continued Research 2.0 Details of the Program 2.1 Marine Coating System Performance	7 8 8 9
Study 2.1.1 Systems Tested 2.1.2 Test Panel Preparation 2.1.3 Test Environment 2.1.4 Evaluation Techniques 2.1.5 Exterior Generic Coating System Test Results	9 9 9 9 10 10
2.1.5.1 Corrosion Protection 2.1.5.2 Chalk Ratings 2.1.5.3 Gloss Results 2.1.5.4 Overall System Performance 2.2 Citric Acid Cleaned Verses Abrasive	20 20 21 21
Blast Cleaned Panels 2.2.1 Primer Test 2.2.1.1 Test Panel Preparation 2.2.1.2 Test Environment and	27 27 27
Evaluation Technique 2.2.1.3 Primer Test Results 2.3 Touch-Up Surface Preparation Test 2.3.1 Test Panel Preparation 2.3.2 Test Results of Touch-Up(Repair) Panels	27 27 33 33 33
2.4 Comparison of Various Generic Types of Primer Used for Touch-Up	35
2.5 Inorganic Zinc Primers Applied Over Four Types of Abrasives	37
References	38

# List of Figures

- Figure 2.1: Vinyl Delamination from Primer
- Figure 2.2: Graphs of Gloss Ratings
- Figure 2.3: Undercutting of Epoxy/Alkyd Coating System
- Figure 2.4: Failure Mode of High Build Polyurethane and Chlorinated Rubber
- Figure 2.5: Checking of Silicone Alkyd
- Figure 2.6: Wash Primer/Polyurethane Failure
- Figure 2.7: Checking of Aromatic Polyurethane
- Figure 2.8: Undercutting of Epoxy/Alkyd Coating System
- Figure 2.9: Touch-Up Panel Prior to Initial Exposure
- Figure 2.10: Touch-Up Panel After 64 Months Exposure

### List of Tables

Table I: Various Generic Coating Systems Exposed on an Exterior

Test Rack (45 Degrees South)

Table II: Summary of Undercutting

Table III: Chalk Evaluation Results

Table IV: Total System Failure Modes

Table V: Citric Acid/Abrasive Blast Performance Summary

Table VI: Various Generic Primers Applied to Abrasive Blast

Cleaned Panels After 66 Months on an Exterior Test Rack

Table VII: Touch-Up Surface Preparation Performance of Various

Primers Applied to Power Tool Cleaned Panels

#### **FOREWORD**

This project was performed under the National Shipbuilding Research Program. The project, as a part of this program, is a cooperative cost shared effort between the Maritime Administration and Avondale Shipyards, Inc. The development work was accomplished by Associated Coatings Consultants under subcontract to Avondale Shipyards, Inc. The overall objective of the program is improved productivity, and therefore, reduced shipbuilding costs.

The studies have been undertaken with this goal in mind, and have followed closely the project outline approved by the Society Of Naval Architects and Marine Engineers' (SNAME) Ship Production Committee.

Mr. Benjamin S. Fultz of Associated Coatings Consultants served as principal investigator. Mr. John Peart of Avondale Shipyards is the R&D Program Manager responsible for technical direction and publication of the final report. Program definition and guidance was provided by the members of the 023-1 Surface Preparation and Coatings Committee of SNAME.

Special thanks are given to the following suppliers for supplying materials and technical direction which made this project possible:

Ameron, Brea California
Byco, Belle Chase, Louisiana
Carboline, St. Louis Missouri
Devoe Marine, Louisville, Kentucky
Farboil, Baltimore, Maryland
Hempel Marine Paints, Houston, Texas
Imperial, New Orleans, Louisiana
International Paint Company, New York, New York
Mobil, Edison, New Jersey
Mobile Paint Manufacturing Company, Mobile, Alabama
Napko, Houston, Texas
Pfizer Inc., Groton, Connecticut
Porter Coating. Louisville, Kentucky
Sherwin-Williams, Cleveland, Ohio
Sigma, New Orleans, Louisiana

#### Executive Summary

The objective of this project was to continue a series of exterior test performance studies which began in 1978 and 1980 as portions of other projects. For a nominal investment, the program has continued for over six years and is now beginning to provide meaningful test results. For the first time, shipyards have access to data which can be used to evaluate the various generic coating systems presently on the market. Even though the state-of-the-art has progressed since the program was initiated, many of the products are still available as originally formulated or with improved formulations. Stated another way, shipyards now have data which can be used to predict actual coatings performance. As an added benefit, accelerated test methods are presented which can be used to screen candidate coating systems.

#### Project Results

# 1.1 Project Overview

This project is a continuation of two performance test programs which began in 1978 and 1980. The first program was entitled "Marine Coatings Performance for Different Ship Areas" and the second was "Cleaning of Steel Assemblies and Shipboard Touch-Up Using Citric Acid". Both programs included accelerated laboratory testing techniques such as Salt Fog Cabinets and Light-and-Water-Exposure Apparatus and exterior Test Fence Exposure (45 Degrees South). This report contains the results of the exterior test fence performance after six years of exposure and attempts to correlate exterior performance with some attributes which can be tested by accelerated laboratory test methods. In addition, various abrasives were used to prepare the substrate of some panels prior to coatings application. Four different types of abrasives were used to prepare panels to which various inorganic zinc primers were applied, and two types were used to prepare the panels to which the generic coating systems were applied. The four abrasives were silica sand, mineral sand, coal slag, and GL-40 steel grit. The two types were mineral sand and GL-40 steel grit.

This report should not be used to qualify, disqualify, compare or select a given supplier or system. The materials used were standard, off-the-shelf materials with no controls exercised to insure that the materials were acceptable prior to use. In addition, no attempt was made to control film thickness to meet manufacturer's recommendations. In some cases, the products tested have been reformulated and/or product designation changed. Some are no longer manufactured or recommended for use as tested. The purpose for presenting the data is to compare general performance of various generic materials and to compare the results to laboratory testing. It must also be remembered that shipyard production influences have not been factored into performance.

The results and conclusions of these programs are as follows:

- 1. Careful selection of laboratory test methods and evaluation parameters, to simulate service conditions, can serve as a screening method for candidate coatings.
- 2. Most generic exterior coating systems continue to provide protection to the steel substrate after 6 1/2 years exposure even though some topcoats have failed.
- 3. The degree of undercutting protection provided by inorganic zinc primer does not appear to be film thickness dependent. Of the 56 systems tested, only 16 had any degree of undercutting. The film thickness of the primers with undercutting and without undercutting varied from 1.8

#### to 5.8 roils.

- 4. More chlorinated rubber systems failed than any other generic type tested. This supports the actual case history analysis of "Marine Coatings Performance for Different Ship Areas" study which found that inorganic zinc with epoxy topcoats outperformed inorganic zinc with chlorinated rubber topcoats.
- 5. Abrasive selection has no measurable impact on overall coating performance.
- 6. Exterior fade and chalk of topcoats roughly correlate with Light-and-Water Exposure Cabinets.
- 7. Salt Fog screening tests can be used for inorganic zinc primer provided the primer is allowed to age in an exterior environment for at least sixty days prior to testing.
- 8. Primers applied over citric acid cleaned steel performed as well as, or superior to, the same primer applied over abrasive blast cleaned steel.
- 9. Of the primers tested, the two component inorganic zinc provides the best corrosion protection.

#### 1.2 Cost Savings

Exact cost savings are difficult to define; however, a properly designed test program can screen proposed candidate paints and identify potentially poor performers. The cost of such a program may seem expensive (approximately \$5,000.00) until it is remembered just how expensive it is to replace the freeboard paint system of a ship at guarantee survey time; 5 to 6 figure range. It must be stressed that any test program be properly designed and controlled. Placing steel plates painted with different materials in the steel storage yard and checking at irregular intervals is not a test program.

#### 1.3 Continued Research

The test fence program should be continued to determine at what point significant generic system or primer failures occur and the steel begins to deteriorate.

The Salt Fog Cabinet and the Light-and-Water Apparatus subject the coating system to different environmental conditions, namely salt spray and ultraviolet/water shock treatments respectively. A test program should be devised to test the synergistic affects of a combination of these effects on a coating system. One approach could be to expose coating systems first in a Light-and-Water Test Apparatus for 200 hours and then in a Salt Fog Cabinet for 100 hours. The test panels would then be cycled between test environments until coating failure. Simultaneously, control

environments until coating failure. Simultaneously, control panels with the same system could be tested in each apparatus without cycling or removal. Results could then be compared.

#### 2.0 Details of the Program

#### 2.1 Marine Coating System Performance Study

This portion of the test program was initially formulated to verify or support actual case histories collected as a part of the original "Marine Coating Performance Study". The exterior freeboard was selected as a representative area. This area was chosen because of the availability of the test environment and the possible potential of collecting adequate numbers of historical data.

#### 2.1.1 Systems Tested

Table I includes the Paint Systems tested. In general, ten suppliers submitted wet samples of paint which were product matches for the generic description of the requested systems. Five primary systems were compared with some alternates being tested. The primer in all cases was a solvent based, (alkyl) inorganic zinc. The topcoats were polyamide epoxy intermediate with and without topcoats of either aliphatic polyurethane, silicone alkyd, or alkyd. The other systems had intermediate and topcoats of either chlorinated rubber or vinyl. The film thicknesses listed are actual film thickness measurements.

# 2.1.2 Test Panel Preparation

The steel panels used for testing were ASTM A-36, 6" X 18" X 1/4" hot rolled plate. All panels were abrasive blasted to Steel Structures Painting Council Surface Preparation Standard, ssPc-SP10, "Near White". Two types of abrasives were used to prepare the panels-mineral sand and steel grit. Some systems were applied over both mineral sand and steel grit prepared substrates and some were only applied over steel grit blasted surfaces. A senior laboratory technician skilled in paint application applied each coating. Material application data sheets supplied by each manufacturer were used to determine thinning, application and overcoat time requirements. No special procedures nor special considerations were granted, and no controls were exercised to precisely control film thickness.

#### 2.1.3 Test Environment

The prepared and painted test panels were exposed on an exterior test rack at 45 South in Jacksonville, Florida less than 100 yards from the St. John's River. The St. John's River at this location has a salt content very similar to the Atlantic ocean which is less than 2 miles away.

## 2.1.4 Evaluation Techniques

Panels were evaluated for rust, chalk, gloss, cracking, blistering and checking using the following ASTM Standards:

Evaluating	the	Degree	of	Rust	ASTM	D610
Evaluating	the	Degree	of	Chalk	ASTM	D659
Evaluating	the	Degree	of	Gloss	ASTM	D523
Evaluating	the	Degree	of	Checking	ASTM	D660
Evaluating	the	Degree	of	Cracking	ASTM	D661
Evaluating	the	Degree	of	Blistering	ASTM	D714

## 2.1.5. Exterior Generic Coating System Test Results

Table I contains the results of these tests, Figures 2.1 and 2.3 thru 2.8 contain photographs of representative test panels. As seen from the test data, differences in chalking and percent change in gloss are easily detected. These results generally agree with other published test results. Epoxies chalk more than chlorinated rubbers and chlorinated rubbers chalk more than aliphatic polyurethane. It can also be seen that in the one case tested, aliphatic polyurethanes outperform aromatic polyurethanes. Most systems continue to provide adequate corrosion protection.

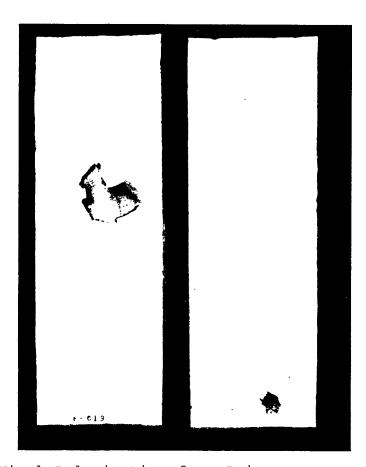


Figure 2.1: Vinyl Delamination from Primer

Table I: Various Generic Coating Systems Exposed On Exterior Test Rack (45 South)

Generic <b>Type</b>	Supplier	Abrasive <b>Type</b>	Prduct No.	Film Thickness	Rating
Inorganic	Ameron	GL-40 Steel Grit	D-6	5.0	(6.5 Yrs.) 10-Rust Gloss Not Evaluated
zinc Synthetic Tiecoat		Steel Gilt	54TC	1.5	Flat Finish
vinyl		Mineral	99	1.1	10-Rust Gloss Not Evaluated
Copalymer vinyl Copolymer		Sand	99	3.6	Flat Finish
Inorganic	Ameron	GL-40	D-6	5.0	10-Rust
Zinc Polyamide		Steel Grit	66	3.0	8-Erosion
Epoxy Polyamide Epoxy		Mineral Sand	66	4.0	10-Rust 8-Erosion
Inorganic	Ameron	GL-40	D-6	4.0	10-Rust
zinc Polyamide		Steel Grit	383	2.5	6-Chalk @ 1 Year 87% Loss in GlOSS @
Epoxy Polyamide Epoxy		Mineral Sand	383	5.5	3 Months 10-Rust 6-Chalk @ 1 Year 87% Loss in GlOSS @ 5 Months
inorganic	Ameron	GL-40	D-6	4.3	10-Rust
Zinc Polyamide	111101 011	Steel Grit	71	1.5	9 .5-Chalk @ 1 Year 50% Loss in Gloss @
Epoxy Silicone			5403	2.6	1 Year
Alkyd Silicone Alkyd		Mineral Sand	5403	1.0	l0-Rust 9-Chalk @ 1 Year 50% Loss in Gloss @ 1 Year
Inorganic	Ameron	GL-40	D-6	4.6	10-Rust
zinc Polyamide Epoxy		Steel Grit	71	1.9	9-Chalk @ 1 Year 46% Loss in Gloss @ 1 Year
Aliphatic Polyurethar Aliphatic Polyurethar		Mineral Sand	2119 2119	1.7 3.7 1/	10-Rust 16"Undercut @ Scribe 9-Chalk @ 1 Year 41% loss in Gloss @ 1 Year

Table I (con't)

Inorqanic	Ameron	G1-40	D-6	5.0	lo-Rust
Zinc Chlorinated Rubber Chlorinated		Steel Grit	2015	2.0	1/16" Undercut @ Scribe 8-chalk @ 1 Year 55% Loss in Gloss @
Rubber Chlorinated		Mineral	2029	1.8	1 Year 10-Rust
Rubber		Sand	2029	3.0	1/16" undercut @ Scribe 8-Chalk 70% Loss in Gloss @ 1 Year
Inorganic Zinc	Carboline	GL-40 Steel Grit	CZll	6.0	10-Rust
Vinyl Copolymer Tiecoat		preel dir	935TC	2.0	6-Chalk 81% Loss in Gloss @ 1 Year. Topcoat Delamination @ 45
vinyl Copolymer		Mineral	938	1.5	Months. See Photo 2.1 10-Rust
vinyl Copolymer		Mineral Sand	938	4.0	6-Chalk 81% Loss in Gloss @ 1 Year. Topcoat Delamination @ 45 Months. See Photo 2.1
Inorganic	Carboline	GL-40	CZll	3.0	10-Rust
zinc Polyamide		Steel Grit	191HB	6.2	8-Chalk 77% Loss in Gloss @
Epoxy Mod. Medium			GP-62	1	1 Year . 8
Oil Alkyd Mod. Medium Oil Alkyd			GP-62	0.8	
Inorganic	Carboline	Mineral	CZll	7.8	10-Rust
Zinc Polyamide		Sand	191НВ	6.2	9-Chalk 30% Loss in Gloss @
Epoxy Aliphatic			132	4.0	1 Year
Polyurethane Aliphatic Polyurethane			132	4.5	
Inorganic	Carboline	GL-40	CZ11	6.0	Failed @ 24 months
zinc Chlorinated		Steel Grit	3630	2.1	6-Chalk @ 1 Year 95% Loss in Gloss @
Rubber Chlorinated		Mineral	3630	0.5	1 Year Failed @ 24 months
Rubber Chlorinated Rubber		Sand	3630	3.0	6-Chalk 95% Loss in Gloss @ 1 Year. See Photo 2.4 (Right Panel #6-024 )

Table I (con't )

<del></del>		OT 40	204	F 0	10 Decemb
Inorganic zinc	Devoe	GL-40 Steel Grit	304	5.0	10-Rust 9-Chalk @ 1 Year
vinyl Tiecoat		50001 0110	MD4368	0.8	6% Loss in GlOSS @ 1 Year. 64% @ 2 Years
vinyl			MD4361	1.0	I Icar. 040 @ Z Icars
Acrylic vinyl Acrylic		Mineral Sand	MD-4361	3.0	10-Rust 1/32" undercut @ scribe 9-Chalk 3% Loss in Gloss @ 1 Year. 60% @ 2 Years
Inorganic	Devoe	GL-40	304	7.0	10-Rust
zinc Polyamide Epoxy		Steel Grit	224	7.8	4-Chalk 4-Erosion 88% Loss in Gloss @ 2 Months. Pinholes From Topcoat Erosion.
Inorganic	Devoe	Mineral	304	6.0	Complete Failure of
Zinc Polyamide		Sand	224	7.0	Topcoat. Cracking/ Alligating. See Photo No. 2.5.
Epoxy Silicone			MD3925	4.0	lo-Rust
Alkyd Silicone Alkyd			MD3925	8.9	
Inorganic Zinc	Eevoe	GL-40 Steel Grit	304	5.0	lo-Rust 8-Chalk @ 1 Year
Polyamide Epoxy			2 2 4	8.0	96% Loss Of Gloss @ 10 14Months.Scme Under-
Acrylic Epoxy		Mineral Sand	229	8.0	cutting @ Scribe & Pinholes from Erosion.
Inorganic zinc	Hempel	GL-40 Steel Grit	1570	3.6	10-Rust 2-Chalk @ 9 Months
Polyamide		beech dire	HB4520	3.0	96% Loss in Gloss @ 4 Months
Fcw? Polyamide Epoxy			5534	3.8	1 MOHOHO
Inorganic zinc	Henpel	GL-40 Steel Grit	1570	3.6	lo-Rust 8-Chalk @ 1 Year
Polyamide		DUCCI OIIC	HB4520	3.5	84% Loss in Gloss @ 7 Months
Epoxy Alkyd			5214	3.5	, PIOTICIES

Table I(con't)

Inorganic Zinc	Hempel	GL-40 Steel Grit	1570	3.6	l0-Rust 9-Chalk @ 1 Year
Polyamide			HB4520	3.8	31% Loss in Gloss @ 1 Year
Epoxy Silicone			5372	3.0	i leai
Aluminum					
(High Heat)					
Inorganic	Imperial	GL-40	555	5.3	lo-Rust
zinc vinyl		Steel Grit	777	3.4	6-Blisters (Few) @ 20 Months
Tiecoat				5.1	8-Chalk @ 1 Year
vinyl			321	3.0	Gloss Not Evaluated,
Topcoat					Flat Finish.
Inorganic Zinc	Imperial	GL-40 Steel Grit	555	5.0	10-Rust 4-Chalk @ 9 Months
Polyamide		Steel Gilt	1200	6.8	1/8" Undercut @ Scribe
Epoxy					Gloss Not Evaluated,
					Flat Finish.
Inorganic	Imperial	GL-40	555	4.2	lo-Rust
zinc Polyamide		Steel Grit	1200	9.6	6-Chalk @ 1 Year 70% Loss in Gloss @
_			1200	J.0	1 Year.
Epoxy Alkyd			88	5.2	
Inorganic	Imperial	GL-40	555	4.5	10-Rust
zinc Polyamide		Steel Grit	1200		8-Chalk @ 1 Year 60% Loss in Gloss @
Epoxy			1200		1 Year, However No
Silicone			84		Change in Gloss for
Alkyd					2nd Year.
Inorganic	Imperial	GL-40	555	4.4	10-Rust
zinc Polyamide		Steel Grit	1200	5.4	9.5-Chalk @ 1 Year 1/16" Undercut @ Scribe
Ероху					19% Loss in Gloss @
Aliphātic Polyurethan	•		1001	2.1	1 Year.
Polyulechan					
Inorganic zinc	Imperial	GL-40 Steel Grit	555	4.7	lo-Rust 8-Chalk @ 1 Year
vinyl		DIECT GIIL	777	2.9	1/8" Undercut @ Scribe
Tiecoat			000	1 ^	49% Loss in Gloss @
Chlorinated Rubber (Acry			890	1.9	1 Year.
. 4	•				

Table I (con't)

Inorganic Internation		2410/11	2.0	lo-Rust 8-Chalk @ 1 Year
Zinc vinyl vinyl Acrylic	Steel Grit	846 3508	1.9 1.5	79% Loss in Gloss @ 1 Year.
vinyl Acrylic	Mineral Sand	3508	1.0	10-Rust 8-Chalk @ 1 Year 77% Loosing Gloss @ 1 Year.
Inorganic Internation		2410/11	2.5	9-Rust 1/4" undercut @ Scribe
Vinyl Wash Primer	Steel Grit	1757/58	1.0	9-Chalk @ 1 Year 69% Loss in Gloss @
Aliphatic Polyurethane		2202/14	2.5	1. Year.
Aliphatic Polyurethane	Mineral Sand	2202/14	3.5	9-Rust. 9-Chalk @ 1 Year 72% Loss in Gloss @ 1 Year. Total Topcoat Delamination @ 5 Years 3 1/2" Underecut. See Photo 2.6.
Inorganic Internation		2410/11	2.3	9-Rust
zinc Vinyl Wash Primer	Steel Grit	1757/58	1.0	4-Checking. See 2.6. 9-Chalk @ 1 Year 40% Loss Gloss @
Aromatic Polyurethane		859	2.5	1 Year.
Aromatic Polyurethane	Mineral Sand	859	2.0	9-Rust 4-Checking 9-Chalk @ 1 Year. 39% Loss in Gloss @ 1 Year.
Inorganic Internation		2410/11	2.0	10 Rust-Pinholes from
zinc Polyamide Epoxy	Steel Grit	8967/ 1539	16.0	erosion of topcoat. 4-Chalk @ 3 Months. 80% Loss in Gloss @ 3 Months.
	Mineral Sand			lo-Rust 4-Chalk @ 4 Months. 87% Loss in Gloss @ 3 Months.
Inorganic Mobile zinc Paint Mfg.	GL-40 Steel Grit	28DH50	1.8	lo-Rust 1/32" Undercut @ Scribe
vinyl vinyl	Dreet Attr	5DR5 5DW2	1.6 2.6	2-Chalk @ 9 Months. Gloss Not Evaluated, Flat Finish.

Table I (con't)

Inorganic	Mobile	GL-40	28DH50	1.6	10-Rust .
Polyamide	Paint Mfg.	Steel Grit	40AH22	6.2	4-Chalk @ 5 Months. Some Checking 91% Loss in Gloss @
Epoxy Polyamide			513-17	2.7	1 Year.
Epoxy					
Inorganic zinc	Mobile Paint Mfg.	GL-40 Steel Grit	28DH50	1.2	10-Rust 9-Chalk @ 1 Year
Polyamide	J		40AH22	6.2	Some Checking 80% Loss in Gloss @
Epoxy Alkyd Tiecoat			28DR105	2.7	1 Year.
Alkyd Topcoat			5010-16	4.1	
Inorganic zinc	Mobile Paint Mfq.	GL-40 Steel Grit	28DH50	1.2	10-Rust Topcoat Delaminated @
Polyamide	raille Mig.	preel Giir	40AH20	6.3	44 Months. Topcoat Applied in Error.
Epoxy Polyvinyl Chloride			5DW2	4.2	Applied in Ellor.
Inorganic	Mobile	GL-40	28DH50	1.1	lo-Rust
Chlorinated	Paint Mfg. d	Steel Grit	548-16	2.0	5-Chalk @ 5 Months Gloss Not Evaluated,
Rubber Chlorinated	d		548-16	3.5	Flat Finish. Some Checking.
Rubber					
Inorganic zinc	Mobil	GL-40 Steel Grit	13F12	2.2	10-Rust 1/4" Undercut @ Scribe
Vinyl		Decer dire	80R8	0.7	4-Chalk @ 1 Year.
vinyl vinyl			83F34 80F34	5.3 3.2	90% Loss in Gloss @ 9 Months.
1		Mineral			10-Rust
		Sand			4-Chalk @ 1 Year. 90% Loss in Gloss @ 9 Months.
Inorganic	Mobil	GL-40	13F12	2.5	lo-Rust
zinc Polyamide		Steel Grit	89F12	6.5	Some Erosion of Topcoat 4-Chalk @ 5 Months.
Epoxy Polyamide			84F34	1.6	90% Loss of Gloss @ 4 Months.
Epoxy		Mineral Sand			10-Rust 4-Chalk @ 5 Months. 91% Loss of Gloss @

Table I (con't)

Inorganic	Mobil	GL-40	13F12	2.5	lo-Rust
Zinc Polyamide		Steel Grit	89F15	9.0	9-Chalk @ 1 Year. 71% Loss in Gloss @
Epoxy Alkyd		Mineral Sand	20F34	1.5	1 Year. lo-Rust 8Chalk @ 1 Year. 68% Loss Gloss @ 1 Year.
Inorganic zinc	Mobil	GL-40 Steel Grit	13F12	2.4	10-Rust 9-Chalk @ 1 Year.
Polyamide		DCCCI GIIC	89F15	9.2	40% Loss in Gloss @ 1 Year.
Epoxy Aliphatic Polyurethane		Mineral Sand	40W9	2.8	10-Rust 8-Chalk @ 1 Year. 40% Loss in Gloss @ 1 Year.
Inorganic	Mobil	GL-40	13F12	2.0	lo-Rust
zinc Polyamide		Steel Grit	89F15	8.3	9-Chalk @ 1 Year. 46% Loss in Gloss @
Epoxy Water Borne Acrylic		Mineral Sand	42F34	1.5	1 Year. lo-Rust 9- Chalk @ 1 Year. 46% Loss in Gloss @
					1 Year.
Inorganic Zinc	Mobil	GL-40 Steel Grit	13F12	2.2	10-Rust,Blistering & Complete Failure of
Chlorinated Rubber		DCCCI GIIC	27F15	4.0	Topcoat @ 56 Months 9-Chalk @ 1 Year.
Chlorinated Rubber		Mineral Sand	28F34	2.8	71% Loss of Gloss @ 1 Year. See Photo 2.8. 10-Rust, No Topcoat Failure.
					8- Chalk @ 1 Year. 70% Loss in Gloss @ 1 Year.
Inorganic zinc	Napko	GL-40 Steel Grit	1375	4.7	10-Rust 1/8" undercut @ Scribe
Copolymer Tiecoat		DUCCI GIIC	1340	1.8	9-Chalk @ 1 Year. Gloss Not Evaluated,
vinyl Topcoat			5452	2.8	Flat Finish.
Inorganic zinc	Napko	GL-40 Steel Grit	1375	4.5	lo-Rust 9-Chalk @ 1 Year.
Vinyl vinyl		DUCCI GIIL	5437 5452	2.3 2.3	Gloss Not Evaluated, Flat Finish.

Table I (con't)

Inorganic	Napko	GL-40	1375	5.5	lo-Rust
zinc Catalyzed Epoxy		Steel Grit	5802	5.2	1/32" Undercut @ Strike 4-Chalk @ 7 Months. 81% Loss in Gloss @ 2 Months.
Inorganic zinc	Napko	GL-40 Steel Grit	1375	4.9	10-Rust. 1/32" Undercut @ Scribe
Polyamide		20001 0110	5616	2.4	8-Chalk @ 1 Year. 90% Loss in Gloss @
Epoxy Alkyd			4318	1.0	9 Months.
Inorganic zinc	Napko	GL-40 Steel Grit	1375	5.8	lo-Rust 1/4" Undercut @ Scribe
Chlorinated Rubber		50001 0110	8-4137	3.0	9-Chalk @ 1 Year. 74% Loss of Gloss @
Chlorinated Rubber			8-4137	2.6	1 Year.
Inorganic zinc	Napko	GL-40 Steel Grit	1375	5.77	lo-Rust 1/4" Undercut @ Scribe
Polyamide		preel diir	5616	1.6	9.5-Chalk @ 1 Year. 15% Loss of Gloss @
Epoxy Polyurethane			5909	2.5	1 Year.
Inorganic zinc	Napko	GL-40 Steel Grit	1375	5.4	Topcoat Delaminated from Inorganic Zinc
High Build Polyurethane			8-4144	3.4	@ 18 Months. 9.5-Chalk @ 1 year.
Polyurethane			5909	3.5	17% Loss of Gloss @ 1 Year. See Photo 2.4 (Left Panel # 6-109)
Inorganic zinc	Porter	GL-40 Steel Grit	351	3.0	lo-Rust 2-chalk @ 9 Months.
Vinyl Wash Primer		preer dir	1799	0.5	Gloss Not Evaluated, Flat Finish.
vinyl			3710	2.0	riac rinish.
Inorganic zinc	Porter	Mineral Sand	351	3.0	lo-Rust 1/32" Undercut @ Scribe
Vinyl Wash Primer		barra	1799	0.5	9.5-Chalk .@ 1 Year. 23% Loss of Gloss @
Aliphatic Polyurethane			4674	2.0	1 Year.
	Sherwin-	GL-40	A6181/B69		10-Rust
zinc High Build vinyl	Williams	Steel Grit	B69A26	(Total D	<pre>1/16" .Undercut @ Scribe   6-Chalk @ 1 Year. FT) Glposs Not Evalu-   ated, Flat Finish.</pre>

Table I (con't)

Inorganic	Sherwin-	GL-40	A6181/B69	lo-rest
zinc Epoxy	Williams	Steel Grit	B69W70	7.7 1/3" Undercut @ Scribe (Total DFT) 4-Chalk @ 7 Months.
1 - 1				91% loss of Gloss @ 2 Months.
				Z MOHCHS.
Inorganic	Sherwin-	GL-40	A6181/B69	7-Rust
zinc	Williams	Steel Grit	B69N70	1 1/2" Undercut @ Scribe 6-Chalk @ 1 Year.
Epoxy Alkyd			B53W10	11.5 89% Loss Of Gloss @
T	Ob o so si so	GL-40	A6181/B69	(Total DFT) 7 Months.See Photo 2.3 9-Rust
Inorganic zinc	Sherwin- Williams	Steel Grit	A0101/B09	1/2" Undercut @ Scribe
Epoxy			B69N70	8-Chalk @ 1 Year.
Aliphatic			F63W13	14 62% Loss of Gloss @
Inorganic	Sherwin-	GL-40	A6181/B69	(Total DFT) 1 Year. lo-Rust
Zinc	Williams		rit	1/32" Undercut @ Scribe
Chlorinated			B69W17	9-Chalk @ 1 Year.
Rubber			D 6 Oti 1 7	67% Loss of Gloss @ 8.5 1 Year.
Chlorinated Rubber			B69W17	8.5 1 Year. (Total DFI)
Modified	Sigma	GL-40	7552	2.3 10-Rust
Inorganic		Steel Grit		Alligating/Pinholes
zinc Polyamide			7430/	56 Months. Complete 5.1 Topcoat Failed @ 66
Epoxy			2190	Months.
Polyamide			7425/	3.6 2-Chalk @ 5 Months.
Epoxy			7000	95% Loss of Gloss @ 5 Months.
Modified	Sigma	GL-40	7552	2.3 lo-Rust
Inorganio	-	Steel Grit		1/32" Undercut @ Scribe
Zinc Polyamide			7430/	9-Chalk @ 1 Year. 6.6 56% Loss Of Gloss @
-			2190	1 Year.
Epoxy Silicone			7238/	0.7
Alkyd	O i arma	OT 40	7000	2 C la Bust
Modified Inorgan	Sigma ic	GL-40 Steel Grit	7552	2.6 lo-Rust 4-Checking
zinc		50001 0110		9.5-Chalk @ 1 Year.
Polyamide			7430/	7.4 7% Loss of Gloss @
Epoxy Aliphatic			2190 7520/	1 Year. 1.9
Polyurethan	e		7000	1.9
Modified	Sigma	GL-40	7552	2.5 lo-Rust
Inorgan zinc	1 C	Steel Grit		4-Checking 8-Chalk @ 1 Year.
Chlorinated			7311/	3.5 60% Loss of Gloss @
Rubber			200	1 Year.
Chlorinated			7310/	3.4
Rubber			200	

#### 2.1 .5.1 Corrosion Protection

With minor exceptions, most of the systems tested continue to provide adequate corrosion protection as concerns ASTM Rust Grades. The primary difference seems to be in the degree of undercutting even though no precise conclusions can be drawn. The following Table summarizes the results:

Table II	:	Summary	of	Undercutting
----------	---	---------	----	--------------

	<u>Undercutting</u>	Percent of Systems With Undercutting
Inorganic Zinc Epoxy	4 of 12 Systems Tested	33%
Inorganic Zinc Epoxy Alkyd	3 of 11 Systems" Tested	27%
Inorganic Zinc Epoxy Polyurethane	4 of 7 Systems Tested	57%
Inorganic Zinc Vinyl	5 of 10 Systems Tested	50%
Inorganic Zinc Chlorinated Rubber	4 of 8 Systems Tested	50%

# 2.1.5.2. Chalk Ratings

Table I contains chalking information. In addition, exterior test results at 6,12 and 18 months compared to the same systems evaluated for 1000 hours in a carbon arc Light-and-Water Apparatus are contained in the following table:

Table III: Chalk Evaluation Results

	est Fence**	1	18 Months	Test Apparatus 1000 Hours
-	6 Months	<u>1 year</u>	16 MOHUHS	<u> 1000 Hours</u>
Epoxy	4.1	4.1	3.9	4.0
Alkyd	8.8	7.9	8.0	8.7
Silicone Alky	d 9.3	9.1	9.1	9.1
Aliphatic Polyurethane	9.5	9.1	8.1	9.4
Vinyl	8.3	6.1	6.2	8.0
Chlorinated Rubber	8.5	7.7	5.4	8.8

<sup>\*</sup>Only finish coats are listed

<sup>\*\*</sup>Average of all systems tested

From these tests, 1,000 hours in the accelerated test chamber appears to approximate six months on the test fence. With most systems, minor change in chalking occurred after six months. The degree of chalking by generic type generally follows the accepted rules for chalking except for the aliphatic polyurethane. Of the materials tested, the silicone alkyd materials outperformed the polyurethane.

#### 2.1.5.3 Gloss Results

Table I presents gloss information as a percent loss of gloss with time. It was necessary to normalize the data in this manner to provide meaningful results because of the wide variance of initial gloss readings. The graphs in Figure 2.2 also compare loss of gloss with time under both accelerated conditions and after exterior test fence exposure. These are selected examples and not averages of all systems tested. One year on the test fence provided reasonable correlation with 1000 hours in the test chamber.

# 2.1.5.4 Overall System Performance

Of the systems tested, the only generic type supplied from two different sources which failed by the same mechanism was the chlorinated rubbers. This may be coincidence; however, the results do somewhat correlate with the original performance study (Reference 3). In that study, chlorinated rubbers did not appear to perform as well as some other generic types. Vinyl wash primer with polyurethanes and high build polyurethane both failed in this test program; however, most suppliers no longer recommend these systems. The epoxy and epoxy/alkyd systems which failed at 66 months may be indicative of the useful life of these generic types; however, numerous other epoxy systems are continuing to perform. Table IV summaries the results of total system failures.

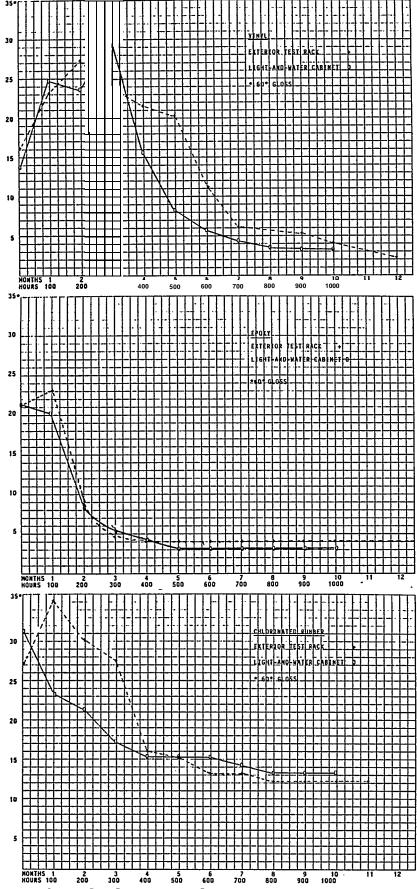


Figure 2.2: Graphs of Gloss Results

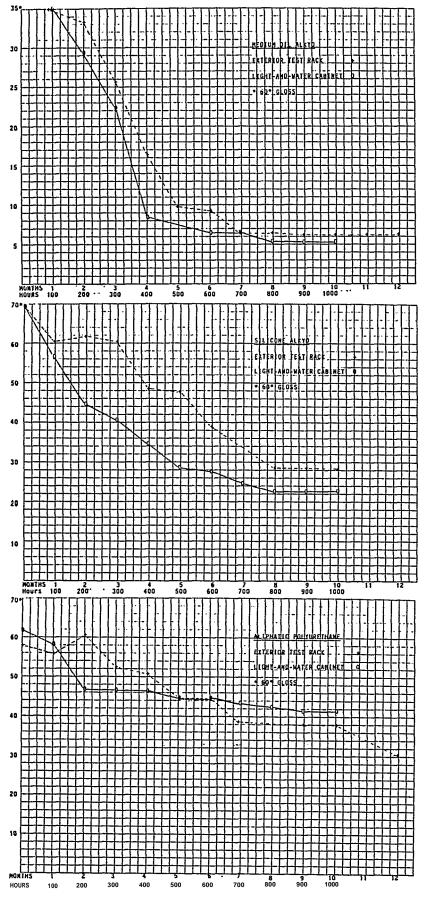


Figure 2.1: (con't)

Table IV: Total System Failure Modes

<u>Generic System</u>	Systems led/Tested	Time	Failure Mode
Epoxy/Silicone Alkyd	lof4	66 months	Checking(Photo 2.5)
Epoxy/Alkyd	lof7	66 months	Delamination from Scribe (Photo 2.3)
Vinyl Wash Primer/ Aliphatic Polyurethane	lofl	60 months	Undercutting from Scribe (Photo 2.6)
High Build Urethane/ Aliphatic Polyurethane	lofl	18 months	Delamination of Topcoat from Primer
Vinyl Wash Primer\ Aromatic Polyurethane	lofl	66 months	Checking(Photo 2.7)
Chlorinated Rubber	_	l@ 24 months l@ 56 months	Topcoat Delamination Complete Topcoat Failure(Photo 2.4 & 2.8)
Ероху	1 of 10	56 months	Alligating\checking
Vinyl	1 o f 9	45 months	Topcoat Delamination (Photo 2.1)

<sup>\*</sup>All systems primed with inorganic zinc.

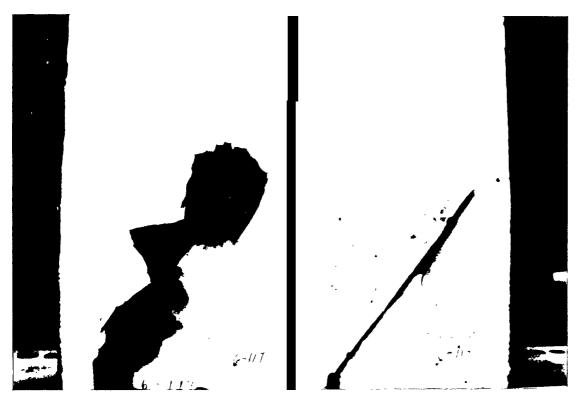
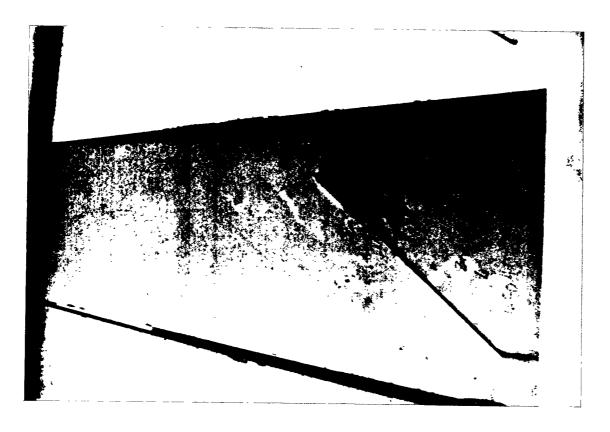


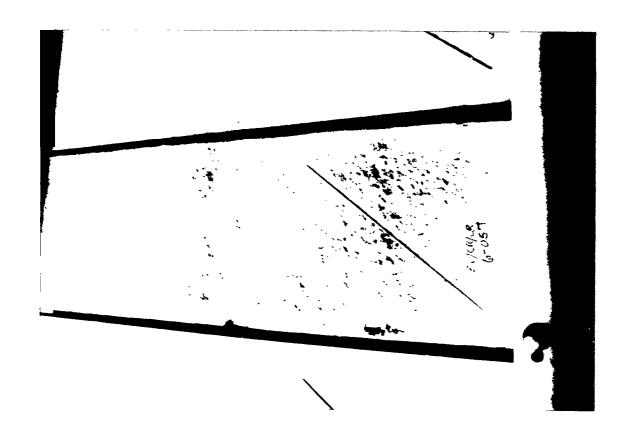
Figure 2.3: Undercutting of Epoxy/Alkyd Coating System

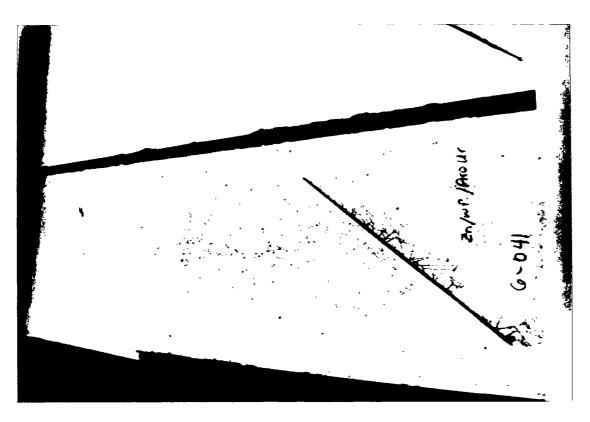


Figure 2.4: Failure Mode of High Build Polyurethane and Chlorinated Rubber Panels









#### 2.2 Citric Acid Cleaned Verses Abrasive Blast Cleaned Panels

There were two different series of exterior test fence exposures of tested primers. The first was a direct comparison of primers applied to both citric acid cleaned panels and abrasive blast cleaned panels. The second was a test to compare citric acid as a touch-up surface preparation technique to the widely used power tool cleaning touch-up technique. The paragraphs which follow discuss each series in detail.

#### 2.2.1 Primer Test

#### 2.2.1.1 Test Panel Preparation

One hundred primers representing seventeen generic types were submitted by ten supplier. Test panels of A-36 steel measuring 6" X 18" were first descaled and then allowed to rust for approximately eight weeks by exposure in an outside industrial, marine environment. Following aged rusting, the panels were divided into two groups. The first group was abrasive blasted to Steel Structures Surface Preparation Standard, SSPC SP 10, "Near White Blast," and the second group was cleaned utilizing a citric acid process. The selected primers were then applied to panels cleaned by each process. Both panels within a set were sprayed at the same time in an effort to duplicate actual film thicknesses. No inhibitors were used with the citric acid process.

### 2.2.1.2 Test Environment and Evaluation Technique

The resulting primed panels were then placed on the test fence at 45 Degrees South for 66 months. Rust grades were determined in accordance with ASTM D610.

#### 2.2.1.3 Primer Test Results

Table VI contains detail application data and performance rating of each primer tested. There were no difference in the performance of post cure inorganic zincs and only minor differences in the water based inorganic zincs applied over both surface preparation methods. The abrasive blasted primers again showed a slightly inferior performance. The remainder of the other types of zinc rich primers also demonstrated almost identical results. Table V contains a summary the results for some of the generic types of primers. As stated earlier no attempt should be made to compare performance between primers of the same generic type and different suppliers or different generic types without taking into account the actual film thickness of the applied materials and the design purpose of each material.

Table V: Citric Acid/Abrasive Blast Performance Summary

<u>Generic Primer</u>	Averaqe R Critric Acid	ust Grade Abrasive Blast
Alkyl Inorganic Zinc	9.6	9.5
One Component Inorganic Zinc	8.2	6.3
Water Based Inorganic Zinc	9.3	8.2
Post Cured Inorganic Zinc	10	9.3
One Component Epoxy Zinc Rich	8.3	7.0
Two Component Epoxy Zinc Rich	8.6	7.0
One Component Epoxy Primer	4.4	3.6
Polyamide Epoxy Primer	5.8	4.9
Polyamine Epoxy Primer	7.5	7.3
Epoxy Ester Primer	7.7	4.5
Alkyd Primer	7.3	6.5
Vinyl Primer	4.7	3.8
Chlorinated Rubber Primer	4.6	5.0

The average performance of all the primers applied over abrasive blasted surfaces was inferior to the performance of those-applied over citric acid. The mean performance of abrasive blast was 6.2, and the mean for citric acid was 7.1. The averaged results are as follows:

2.6	ABRASIVE BLAST	<u>CITRIC ACID</u>
Mean	6.2	7.1
Standard	Deviation 3.8	3.4
Variance	14.3	11.6

Various Generic Primers Applied to Abrasive Blast Cleaned and Citric Acid Cleaned Panels After 66 Months Exposure On Exterior Test Rack (45 Degrees)

			<u>.</u>			
GENERIC		SUPPLIER	PRODUCT	SURFACE	FILM	RUST
TYPE				PREPARATION	THICKNESS	GRADE
Alkyd Inorganic	Zinc	Ameron	D-9	Abrasive Blas		9
Solvent Base				Citric Acid	4.8	9
Alkyd Inorganic	Zinc	Вусо	101	Abrasive Blas		8
Solvent Base				Citric Acid	2.4	88
Alkyd Inorganic	Zinc	Carboline	CZ11	Abrasive Blas		10
Solvent Base			· · · · · · · · · · · · · · · · · · ·	<u> Citric Acid</u>	4.2	10
Alkyd Inorganic	Zinc	Carboline	CWll	<u>Abrasive Blas</u>	st 1.6	Failed 32 Mo
Solvent Base				<u> Citric Acid</u>	1.4	10 @ 32 Mo
Alkyd Inorganic	Zinc	Devoe	304	<u>Abrasive Blas</u>		10
Solvent Base				Citric Acid	2.6	· 10
Alkyd Inorganic	Zinc	Farboil	114	Abrasive Blas	st 3.0	9
Solvent Base				Citric Acid	2.7	9
Alkyd Inorganic	Zinc	Imperial	555	Abrasive Blas	st 3.0	10
Solvent Base		-		Citric Acid	2.7	10
Alkyd Inorganic	Zinc	Internation	al QHAO27	/ Abrasive Blas	st 4.6	10
Solvent Base			QHAO28		4.7	10
ALKYD Inorganic	Zinc	Mobil	13F12	Abrasive Blas		10
Solvent Base				Citric Acid	1.6	10
Alkyd Inorganic	Zinc	Napko	1375	Abrasive Blas		9
Solvent Base '		*		Citric Acid	4.2	10
Alkyd Inorganic	Zinc	Porter	351	Abrasive Blas		10
Solvent Base				Citric Acid	2.1	10
Modified Alkyd		Devoe	302R	Abrasive Blas		6
Inorganic Zinc				Citric Acid	3.0	8
One Component		Ameron	160	Abrasive Blas		9
Inorganic Zinc				Citric Acid	3.2	9
One Component		Ameron	2155	Abrasive Blas		6
Inorganic Zinc		11.02.011	<b>2</b> 233	Citric Acid	3.6	9
One Component		Вусо	102SP9			9
Inorganic Zinc		Dy CC	102019	Citric Acid	6.5	9
One Component	· · · · ·	Devoe	306	Abrasive Blas		4
Inorganic Zinc		Devoe	300	Citric Acid	4.0	9
One Component		Devoe	308	Abrasive Blas		Failed 18 Mo
Inorganic Zinc		Devoc	500	Citric Acid	1.4	8 @ 18 Mo
One Component		Devoe	309	Abrasive Blas		9
Inorganic Zinc		DCVCC	303	Citric Acid	2.0	9
One Component		Imperial	545			10
Inorganic Zinc		пиреттат	: 343	Citric Acid	3:6	10
		Internationa	1 NOA200			
One Component	_	mineriaciona	T INCHZOO			Failed 18 Mo
Inorganic Zinc		Mob: 1	12010	Citric Acid		Failed 18 Mo
One Component		Mobil	13G10	Abrasive Blas		7
Inorganic Zinc		371	. 1203	Citric Acid	2.4	10
One Component		Napko	1301	Abrasive Blas		9
Inorganic Zinc	1.5			Citric Acid	5.4	9
Water Based, Se		Ameron	D-4	Abrasive Blas		10
Cure, Inorganic	Zinc			Citric Acid	4.1	10

Table VI (cont'd)

GENERIC	SUPPLIER	PRODUCT	SURFACE	FILM	RUST
TYPE	SOFFITER	NO.	PREPARATION	THICKNESS	GRADE
Water Based, Self	Devoe	305	Abrasive Blast	4.3	9
Cure, Inorganic Zine		303	Citric Acid	3.5	10
Water Based, Self	Farboil	76	Abrasive Blast	5.0	10
Cure, Inorganic Zine		70	Citric Acid	4.5	10
Water Based, Self	International	100ΔΩΤ	Abrasive Blast	3.1	10
Cure, Inorganic Zin		TQA001/	Citric Acid	3.0	10
Water Based, Self	Mobil	46Fl	Abrasive Blast	4.3 Fail	
Cure, Inorganic Zin		1011	Citric Acid	3.8	6
Water Based, Self	Napko	1371	Abrasive Blast	5.1	10
Cure, Inorganic Zine	_	1571	Citric Acid	5.3	10
Post Cure,	Ameron	D-3	Abrasive Blast	4.6	10
Inorganic Zinc	PHICLOIT	ЪЗ	Citric Acid	4.3	10
Post Cure,	Napko	1361	Abrasive Blast	3.3	10
Inorganic Zinc	rapro	1301	Citric Acid	3.1	10
One Component	Byco 1	50-1	Abrasive Blast	4.1	8
Epoxy Zinc Rich	Dyco 1	.50 I	Citric Acid	3.6	10
One Component	Imperial	512	Abrasive Blast	3.6	8
Epoxy Zinc Rich	nuferrar	J12	Citric Acid	2.9	9
	nternational E	ו א א אי	Abrasive Blast	3.0 Fail	
Epoxy Zinc Rich	ilcernacional El	.V44T	Citric Acid		@ 3 Mo
One Component	Mobil 518	F208	Abrasive Blast	4.0	10
Epoxy Zinc Rich	robit 510	F 200	Citric Acid	2.9	10
One Component	Napko	1355	Abrasive Blast	9.4	7
Epoxy Zinc Rich	тарло	1000	Citric Acid	9.2	9
One Component	Porter	309	Abrasive Blast	3.4	10
Epoxy Zinc Rich	101001	303	Citric Acid	3.3	10
Two Component	Byco ]	.50-5	Abrasive Blast	4.5	9
Epoxy Zinc Rich	Dyco 1	.50-5	Citric Acid	4.3	9
Two Component	Farboil	28	Abrasive Blast		ed 32 Mo
Epoxy Zinc Rich	rationi	20	Citric Acid	2.4 Fall	5 5
Two Component	Mobil	13F4	Abrasive Blast	2.4	7
Epoxy Zinc Rich	TICOM	1314	Citric Acid	2.3	
Two Component	Manko	5614		<u> </u>	9
<del>-</del>	Napko	2014	Abrasive Blast	5.4	10
Epoxy Zinc Rich	Porter	308	Citric Acid Abrasive Blast	3.8	10
Two Component Epoxy Zinc Rich	LOT CET	200	Citric Acid	3.6	
Organic Zinc,	Przec	.50-7	· · · · · · · · · · · · · · · · · · ·	3.7	8
	Byco ]	.50-7	Abrasive Blast		
Chlorinated Rubber	Enrhoil 70	(Mil	Citric Acid	3.7	8
Organic Zinc		(Mil-	Abrasive Blast	3.9	9
One Component		<u>1048)</u>	Citric Acid	3.9	9
One Component	Ameron	185	Abrasive Blast	2.9	8
Epoxy Primer		50.0	Citric Acid	2.7	10
One Component	Byco ]	.50-2	Abrasive Blast	1.7 Fail	
Epoxy Primer		70546	Citric Acid	1.2 Fail	
One Component	Farboil l	2546	Abrasive Blast	1.7 Fail	
Epoxy Primer	<del>-</del> • • •	1015	Citric Acid	1.3 Fail	
One Component	Imperial	1215	Abrasive Blast		<u>ed 13 Mo</u>
Epoxy Primer		77.000	Citric Acid		@ 13 Mo
-	nternational NI	2A2UU	Abrasive Blast	2.8	8
Epoxy Primer			Citric Acid	2.6	8

# TABLE VI(con't)

Citric Acid	On a Company	Nania	1340	Abrending Dlast	2.6	10
Polyamide	One Component	Napko	1340	Abrasive Blast	2.6	10
Polyamide		Jmowon	71	<del></del>		
Polyamide	-	Alleron	11			
Proxy	Epoxy	011	- 102			
Polyamide		Carpolin	e 193			
Poxy	Epoxy					
Polyamide		Devoe	202			
Popoxy	Epoxy					
Delyamide   Devoe   230FD	Polyamide	Devoe	208			
Epoxy	Epoxy					
Epoxy	Polyamide	Devoe	230FD			
Epoxy	Epoxy			Citric Acid		
Polyamide	Polyamide	Farboil	4202			
EDOXY	Epoxy			Citric Acid		5 @ 13 Mo
EDOXY	Polyamide	Farboil	NAVY	Abrasive Blast		7
Epoxy	Epoxy	F	or. 150	Citric Acid	3.4	8
Epoxy	Polyamide	Imperial	1219	Abrasive Blast		9
Polyamide	_	-		Citric Acid	5.3	10
Epoxy		International	EPA0061\	Abrasive-Blast	3.9	Failed 32 Mo
Polyamide	_		EBA744		3.7	7 @ 32 Mo
Epoxy		Mobil	65T1\	Abrasive-Blast		
Polyamide   Porter   Aspect   Polyamide   Porter   Aspect   Aspect   Polyamide   Porter   Porter   Polyamide   Poly	<b>-</b>		•			
Citric Acid   2.2   6		Napko				
Polyamide	••					
MCR43		Porter	4300			
Polyamide						
Polyamine		Porter				
Polyamine	_					
Polyamine		Ameron	2156			
Polyamine						
Figure   Farboil   Farbo		Byco	E-Prime			
Polyamine   Carboline   187HFP   Abrasive Blast   7.0   6		-1				
Citric Acid   7.6   7		Carboline				
Polyamine	<del></del>	0022022110	10.1111			
Time		Mohil	71F84B\			
Polyamine		PODII	•			
Polyamine		Mohil 1				
Polyamine			•			
Citric Acid   3.5   10						
Polyamine         Porter         7650         Abrasive Blast         2.0         6 @ 7 Mo           Epoxy         Epoxy Ester         Byco         360-1         Abrasive Blast         3.2         9           Citric Acid         3.1         9           Epoxy Ester         Farboil         8229         Abrasive Blast         1.8 Failed-32 Mo           Citric Acid         2.2         6 @ 32 Mo           Alkyd         Byco         400-2         Abrasive Blast         2.5         7           Citric Acid         2.5         8           Alkyd         Farboil         1253         Abrasive Blast         3.3         7           Citric Acid         3.0         8           Alkyd         Farboil         6031         Abrasive Blast         2.3         7	_	Mapro	3020			
Citric Acid   1.8   7 @ 7 Mo		Dorehore	7650			
Epoxy Ester         Byco         360-1         Abrasive Blast Citric Acid         3.2         9           Epoxy Ester         Farboil         8229         Abrasive Blast Citric Acid         1.8 Failed 32 Mo Citric Acid         2.2         6 @ 32 Mo Alkyd           Alkyd         Byco         400-2         Abrasive Blast Acid         2.5         7           Citric Acid         2.5         8           Alkyd         Farboil         1253         Abrasive Blast Acid         3.3         7           Citric Acid         3.0         8           Alkyd         Farboil         6031         Abrasive Blast Bl		Porcer	7650			
Citric Acid   3.1   9		D	260.1			
Epoxy Ester         Farboil         8229         Abrasive Blast Citric Acid         1.8 Failed 32 Mo Citric Acid         2.2 6 @ 32 Mo Representation Acid         2.5 7 Representation Acid         7 Representation Acid         2.5 8 Representation Acid         9 Re	Epoxy Ester	вусо	360-1			
Citric Acid         2.2         6 @ 32 Mo           Alkyd         Byco         400-2         Abrasive Blast         2.5         7           Citric Acid         2.5         8           Alkyd         Farboil         1253         Abrasive Blast         3.3         7           Citric Acid         3.0         8           Alkyd         Farboil         6031         Abrasive Blast         2.3         7	Daniel Dalana	D111	0000	<del></del>		
Alkyd         Byco         400-2         Abrasive Blast         2.5         7           Citric Acid         2.5         8           Alkyd         Farboil         1253         Abrasive Blast         3.3         7           Citric Acid         3.0         8           Alkyd         Farboil         6031         Abrasive Blast         2.3         7	Epoxy Ester	rarboll	8229			
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	2117	D	400.0			
Alkyd         Farboil         1253         Abrasive Blast         3.3         7           Citric Acid         3.0         8           Alkyd         Farboil         6031         Abrasive Blast         2.3         7	Alkya	RACO	400-2			
Alkyd Farboil 6031 Abrasive Blast 2.3 7	- 7 7		3.000			<u> 8</u>
Alkyd Farboil 6031 Abrasive Blast 2.3 7	ALKYC	Farboil	1253			
•						
Citric Acid 2.1 8	Alkyd	Farboil	6031			
				Citric Acid	2.1	88

Table VI (cont'd.)

GENERIC	SUPPLIEF	R PRODUCT	SURFACE	FIIM	RUST
TYPE		NO.	PREPARATION	THICKNESS	GRADE
Alkyd	Imperial	62	Abrasive Blast	2.9	8
			Citric Acid	2.7	8
Alkyd	International	CPA476	Abrasive Blast	2.4	6
			Citric Acid	2.2	7
Alkyd	Mobil	53R1	Abrasive Blast	2.8	3
			Citric Acid	2.8	3
Alkyd	Napko	1313	Abrasive Blast	2.7	7
			Citric Acid	3.0	9
Alkyd	Porter	297	Abrasive Blast	2.5	7
		<u>.</u>	Citric Acid	2.6	7
Vinyl	Ameron	86	Abrasive Blast	1.6 Faile	ed 4 Mo
			Citric Acid	1.0 Faile	ed 4 Mo
Vinyl	Ameron	33	Abrasive Blast	2.4 Faile	ed 7 Mo
			Citric Acid	2.0 Faile	ed 7 Mo
Vinyl	Вусо	600-2	Abrasive Blast	2.2	7
	_		Citric Acid	1.7	7
Vinyl	Carboline	8HB	Abrasive Blast	2.8 Faile	ed 32 Mo
•		`	Citric Acid	2.9 6	
Vinyl	Farboil	6600S	Abrasive Blast	3.2	6
_			Citric Acid	3.1	5
Vinyl	International	VXL000	Abrasive Blast	3.3	10
4			Citric Acid	3.0	10
Vinyl Wash Primer	Porter	VC17	Abrasive Blast	1.2 Faile	ed 3 Mo
4			Citric Acid	0.9 Faile	ed 3 Mo
Chlorinated	Carboline	3631	Abrasive Blast	2.3	6
Rubber			Citric Acid	2.4	6
Chlorinated	Devoe	MD3500	Abrasive Blast	1.7 Faile	ed 13 Mo
Rubber			Citric Acid	1.6 Faile	
Chlorinated	Farboil	58ACG	Abrasive Blast	1.9 Faile	
Rubber			Citric Acid	1.6 Faile	
Chlorinated	Imperial	880	Abrasive Blast	4.8	7
Rubber	*		Citric Acid	5.0	6
Chlorinated	International	LPA300	Abrasive Blast	2.8	4
Rubber			Citric Acid	2.8	4
Chlorinated	Mobil	67F34	Abrasive Blast	3.9	9
Rubber			Citric Acid	4.2	9
Chlorinated	Napko	5202	Abrasive Blast	4.2	6
Rubber	<u></u>	<del></del>	Citric Acid	4.1	6
Ketamine	Devoe	244HS	Abrasive Blast	3.7	8
Epoxy	20.00		Citric Acid	3.3	6
Bituminous	Devoe	4314	Abrasive Blast	2.5 Faile	
	20.00		Citric Acid	2.3 Faile	od 13 Mo
Bituminous	International	JAA021	Abrasive Blast	3.8	10
	111001124 01101101	0111021	Citric Acid	3.6	10
Phenolic-Vinyl	International	NFA081	Abrasive Blast	2.1	8
			Citric Acid	2.1	8
Water Borne	Вусо	500-1	Abrasive Blast	2.4 Faile	
(Emulsion)	TYCO	200 T	Citric Acid	2.4 Faile	
Water Borne	Farboil	8285	Abrasive Blast	3.1 Faile	
(Emulsion)	Fariori	0203	Citric Acid	3.1 Faile	
/TERMISTOIL/			CTITTO MOTO	2.T Latte	54 32 MO

#### 2.3 Touch-Up Surface Preparation Test

#### 2.3.1 Test Panel Preparation

Twenty different primers representing twelve generic types were selected at random for the touch-up surface preparation test. The test panels were 6" X 18", A-36 steel panels which were first abrasive blasted to Steel Structure Painting Council Surface Preparation Standard SSPC SP 10, "Near White Blast" and then primed. Each primer selected was applied to the top and bottom third of two each, steel panels. The center third was left bare. Following cure of each coating, a 3/4" weld was made through a portion of the coating and into the unpainted area. See Figures 2.9 for an example of a panel prior to exposure. The prepared panels were then placed on an exterior test rack at 45 South for ten weeks and allowed to rust. After the exposure period, panels were removed from the rack and one panel from each set was touch-up cleaned using a citric acid spray technique, panel from each set was power tool cleaned in accordance with the procedure defined for erection joints in "Catalog of Existing Small Tools for Surface Preparation and Support Equipment for Blasters and Painters." During the citric acid operation it was noted that the citric acid reacted with the alkyl inorganic zinc types of primers (solvent based) and removed the majority of the zinc leaving the panel essentially bare. The water based self cure was removed to a lesser degree and the post cure inorganic It must also be pointed out zinc was not disturbed. that the citric process did not remove residual weld slag or heat damaged initial primer. No attempt was made to supplement the citric acid cleaning with mechanical cleaning prior to touch-up priming. touched-up panels were preprimed and placed back on the exterior test fence at 45 South for 64 months.

### 2.3.2 Test Results of Touch-Up (Repair) Panels

Table VII contains a tabulation of the test results. The overall performance of the citric acid touch-up cleaned surfaces was inferior to the power tool touch-up cleaned surfaces. Figure 2.10 also shows a direct comparison of the performance of power tool cleaning and citric acid cleaning(citric acid panels are on the right in each panel set). The citric acid cleaned primer failure is due to weld damaged paint. In conclusion, citric acid cleaning for touch-up of damaged weld areas must be supplemented with a mechanical cleaning method to remove residual slag, weld splatter, and damaged paint.

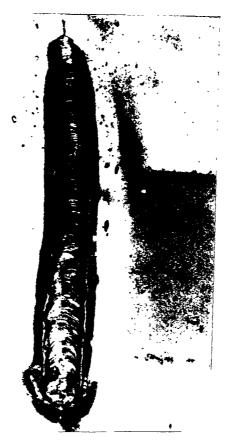


Figure 2.9: Touch-Up Panel Prior to Initial Exposure

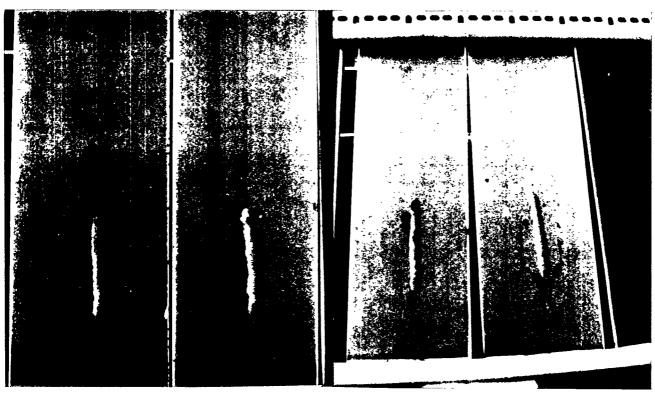


Figure 2.10: Touch-Up Panels After 64 Months Exposure

# 2.4 Comparison of Various Generic Types of Primers

In addition to the observations concerning the comparison abrasive blast panels and citric acid cleaned panels, between other comparisons of generic types can be drawn. For example, the two component inorganic zincs outperformed all other exposed on the test fence. With the exception of one sed, self cured product which failed at three months, water based . the remainder continue to provide excellent corrosion protection. It can also be noted that, of the systems tested, the two zinc primers outperformed the organic zinc component inorganic rich materials. Another interesting finding concerns the one component inorganic zinc primers applied over abrasive blast cleaned panels. Two failed at 18 months with two others having a rust grade rating of 4 and 6 respectively at 66 months. The alkyd are good performers, surpassing the polyamides epoxies, primers and chlorinated rubbers. The one component epoxy is the vinyls worst performer of those tested after 66 months; however, these materials are only designed for 6 to 9 months protection prior to topcoating. It should also be noted that one aluminum pigmented bituminous primer applied 3.8 rolls dry has no rust.

Table VII: Touch-up Surface Preparation Performance of Various Primers Applied to Either Power Tool Cleaned or Citric Acid Cleaned Prepared Panels After 64 Months

GENERIC	SUPPLIE	R PRODUCT	SURFACE	FILM	RUST
TYPE		NO.	PREPARATION	THICKNESS	GRADE
Post Cure	Ameron	D-3	Power Tool	5.6	9
Inorganic Zinc			Citric Acid	5.3	10
Water Based, Self	Ameron	D4	Power Tool	2.5	10
Cure Inorganic Zinc			Citric Acid	2.1	10
Alkyd Inorganic	Carboline	∋ CZll	Power Tool	4.8	10
Zinc			Citric Acid	4.3	10
Alkyd Inorganic	Mobil	13F12	Power Tool	3.3	10
Zinc			Citric Acid	2.7	10
Alkyd Inorganic	Sigma	711G	Power Tool	4.0	9
Zinc			Citric Acid	3.4	9
Alkyd Inorganic	Mobil	28DH50	Power Tool	2.3	9
Zinc ·			Citric Acid	1.8	9
One Component	Devoe	306	Power Tool	5.6	9
Inorganic Zinc			Citric Acid	4.6	10
One Component	Mobil	13G10	Power Tool	2.2	Note 1
Inorganic Zinc			Citric Acid	1.6	Note 2
Modified	Porter	352	Power Tool	3.0	10
Inorganic Zinc			Citric Acid	2.5	10
One Component	Napko	1355	Power Tool	5.6	9
Epoxy Zinc Rich			Citric Acid	4.5	9
Polyamide	Carboline	193HB	Power Tool	5.6	10
Epoxy			Citric Acid	4.3	10
Polyamide	Devoe	208	Power Tool	2.4 I	Failed 30 Mo
Epoxy			Citric Acid	2.0 I	Failed 30 Mo
Polyamide	Napko	5616	Power Tool	2.4	9
Epoxy			Citric Acid	7.0	8
Alkyd	Imperial	62	Power Tool	4.7	8
			Citric Acid	5.4	8
One Component	INT	NEA200	Power Tool	3.4	10
Epoxy			Citric Acid	3.3	9
Ketamine	INT	TTA424	Power Tool	5.9	Note 3
Epoxy			Citric Acid	5.8	8

Note 1: Failed in Repair Area Note 2: Failed in Top Half of Panel, Repair Area Rust Grade 10 Note 3: Failed in Weld Area

# 2.5 Inorganic Zinc Primers Applied Over Four Types of Abrasives

To investigate the possible impact of abrasive selection on paint performance, a limited test program was initiated to test the performance of inorganic zinc primers applied over four different abrasives. Six alkyl inorganic zinc primers were applied to two sets of panels prepared using a coal slag, a mineral sand, a silica sand, and GL-40 steel grit abrasives. Film thicknesses within a supplier set were controlled by applying the materials to all four panels simultaneously. Film thicknesses between supplier sets ranged from 2.3 to 7.0 roils. All panels were then exposed on an exterior test rack. After 60 days, one set was removed and placed in a salt fog cabinet for 6000 hours. The salt fog test was performed in accordance with ASTM B117. After 6000 hours, all panels had a rust grade of 10. In addition, all panels which were left exposed on the test fence for 66 months, within a supplier set, had the same degree of rust. Rating between sets varied from 9 to 10 rust grades.

#### References

- 1. Peart, John, "Catalog of Existing Small Tools for Surface Preparation and Support Equipment for Blasters and Painters", The National Shipbuilding Research Program, May, 1977.
- 2. Fultz, Benjamin S. "Cleaning of Steel Assemblies and Shipboard Touch-Up Using'Citric Acid (Phase I), The National Shipbuilding Research Program", May, 1980.
- 3. Fultz, Benjamin S., "Marine Coating Performance for Different Ship Areas", The National Shipbuilding Research Program, July 1979.